

Density-dependent Transport and Sequential Biotransformation of Trichloroethylene in a Variably Saturated Zone

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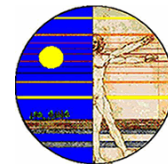
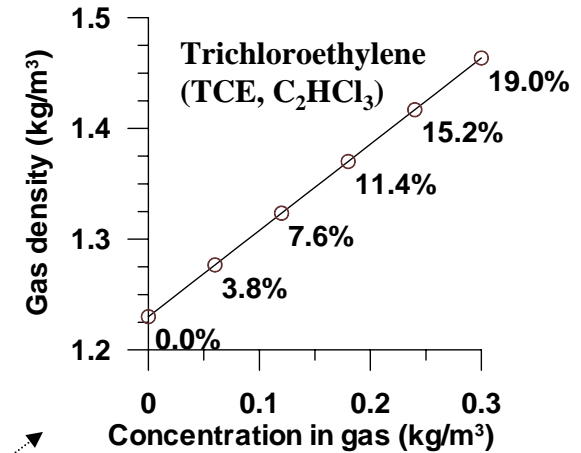


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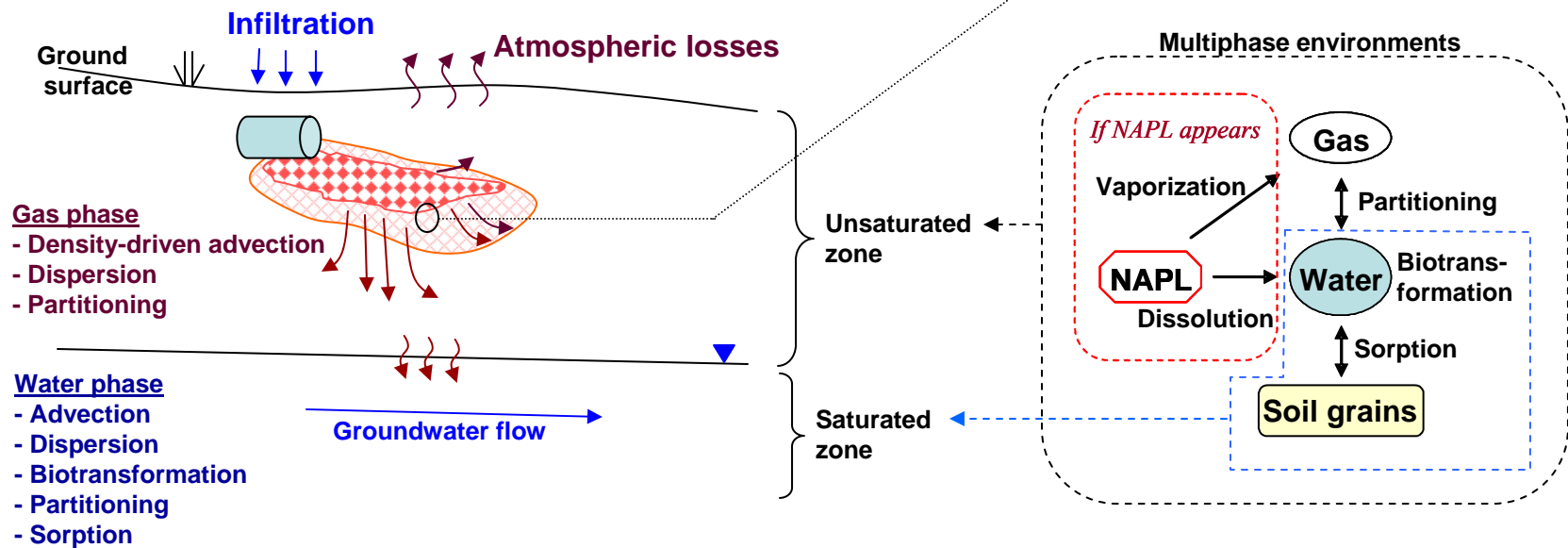
- **Introduction**
- **Study objectives**
- **Governing equations**
 - Flow equations
 - Transport equations
- **Model verification**
- **Density-driven flow and transport**
 - Simulation and scenarios
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Introduction

- **Density-driven advection of gas phase**
 - Is generated by density-gradient within gas phase.
 - Occurs near contaminant source zones.



$$\rho_g = f(P_g, C_g)$$



NAPL = Non-Aqueous Phase Liquid

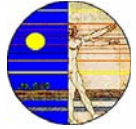


Study objectives

- **Objective is:**
 - To investigate density-driven transport of multi-species with biological reactions in a variably saturated subsurface.

- **Model development activities:**
 - Develop a three-dimensional numerical model
 - Verify and validate the model using analytical solutions, experimental data, and numerical results available in literature

- **Assumptions:**
 - First-order relations
: Biotransformation, sorption, and partitioning
 - Gas phase: $\rho = f(P, C)$ and $\mu = f(C)$
 - Water phase: Constant properties
 - NAPL: Immobile residuals



Flow equations

- From mass conservation and continuity equations

$$\frac{\partial(\phi s_f \rho_f)}{\partial t} - \nabla \cdot \left\{ \rho_f \underbrace{\frac{\mathbf{k}_m k_{rf}}{\mu_f} \cdot [\nabla(\psi_f \rho_w \mathbf{g}) - \rho_f \mathbf{g}]}_{q_f, \text{Darcy velocity}} \right\} = I_f + \rho_f^* Q_f$$

Subscript f = fluid phases
(water and gas)

ψ_f = Pressure head (primary variable)

s_f = Saturation

k_{rf} = Relative permeability

ρ_f = Density

- Gas density

$$\rho_g(P_g, C) = \rho_{air} + \gamma_g P_g + \sum_{i=1}^N C_g^i \left(1 - \frac{\rho_{air}}{\rho_v^i} \right) \quad \begin{array}{l} i = \text{contaminants} \\ N = \text{total number of contaminants} \end{array} \quad (\text{Thomson et al., 1997})$$

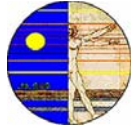
- Relative permeability = f (effective saturation)

– Water phase $k_{rw} = s_{we}^{1/2} \left\{ 1 - \left[1 - s_{we}^{1/m} \right]^m \right\}^2$ (*van Genuchten, 1980*)

– Gas phase $k_{rg} = s_{ge}^{1/2} \left[1 - (1 - s_{ge})^{1/m} \right]^{2m}$ (*Parker et al., 1987*)

- Effective saturation $s_{we} = \left[1 + (\alpha_{gw} \psi_{gw})^n \right]^{-m}$ α_{gw} , n = experimental coefficient
- $$s_{ge} = 1 - s_{we} - s_{ne} \quad \text{m} = 1 - 1/n$$

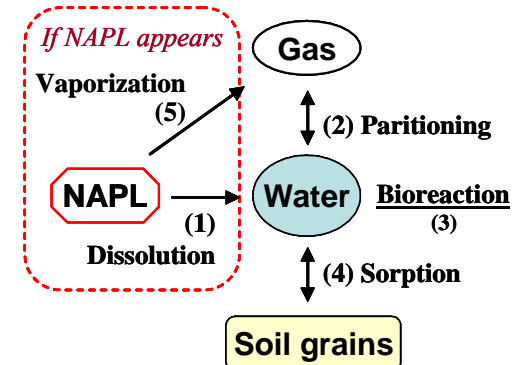
- Gas viscosity = f (concentration) as gas mixture (*Wilke equation, Reid et al., 1987*)



Contaminant Transport Equations

- Multi-species in water and gas phases

$$\frac{\partial(\phi s_f C_f^i)}{\partial t} = \underbrace{\nabla(\phi s_f D_f^i \nabla C_f^i)}_{\text{Dispersion}} - \underbrace{\nabla(q_f C_f^i)}_{\text{Advection}} + \underbrace{I_f^i}_{\text{Mass transfer / Bioreaction}} + \underbrace{Q_f C_f^{*i}}_{\text{Pumping, Injection}}$$



- Contaminant (ith) in water phase

$$I_w^i = \underbrace{\phi s_w \lambda_D^i (C_{we}^i - C_w^i)}_{(1) \text{Dissolution}} + \underbrace{\phi s_g \lambda_H^i (C_g^i - H^i C_w^i)}_{(2) \text{Partitioning : water-gas phase}} + \underbrace{\phi s_w \lambda_B^{i-1} Y_{i-1/i} C_w^{i-1}}_{(3) \text{Generation by biodegradation of parent contaminant, } i-1} - \underbrace{\phi s_w \lambda_B^i C_w^i}_{(3) \text{Biodegradation of contaminant, } i} - \underbrace{\rho_b K_D^i \frac{\partial C_w^i}{\partial t}}_{(4) \text{Sorption : water-soil}}$$

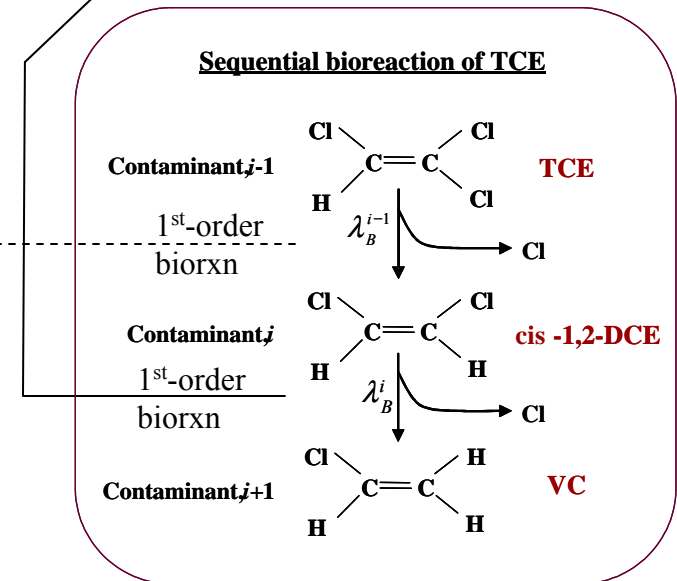
- Contaminant (ith) in gas phase

$$I_g^i = \underbrace{\phi s_g \lambda_V^i (C_{ge}^i - C_g^i)}_{(5) \text{Vaporization from NAPL}} - \underbrace{\phi s_g \lambda_H^i (C_g^i - H^i C_w^i)}_{(2) \text{Partitioning : water-gas}}$$

NAPL saturation

$$\frac{\partial}{\partial t}(\rho_n \phi s_n) = \underbrace{-\phi s_w \lambda_D (C_{we} - C_w)}_{(1) \text{Dissolution}} - \underbrace{-\phi s_g \lambda_V (C_{ge} - C_g)}_{(5) \text{Vaporization}}$$

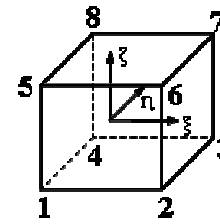
- First-order relation coefficients : $\lambda_D, \lambda_H, \lambda_B, \lambda_V$
- Yield coefficient = DCE mw / TCE mw



Numerical method

- **Galerkin Finite Element Method (FEM)**

- Modified Picard method
- Element of domain
 - Rectangular prism
 - 8 nodes each element
- Three-dimensional mesh generator



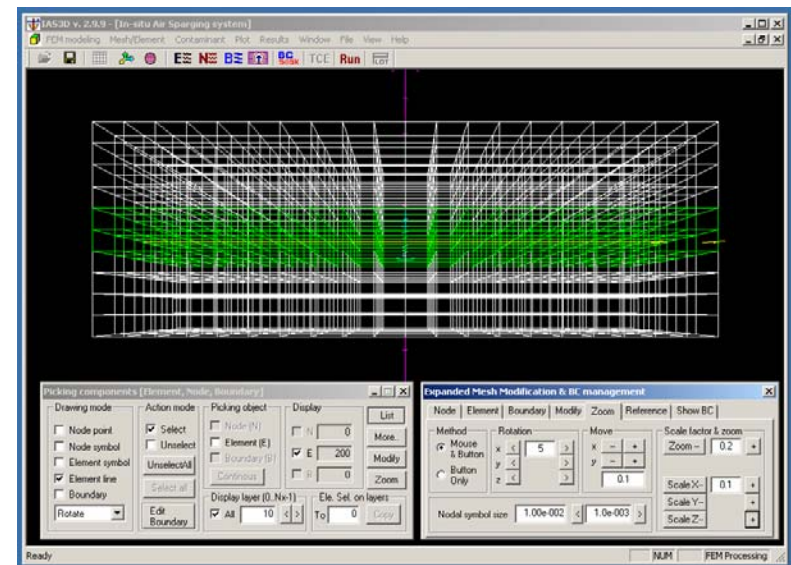
Element with eight nodes

- **Material balance calculation**

- Accuracy and error checking

- **Numerical codes (TechFlow^{MP})**

- Program language
 - C, C++, and Visual C++
- Support platform
 - Linux
 - Unix (High Performance Computing)
 - Microsoft Window



TechFlow^{MP}
Graphical user interface and 3D mesh generator



Model verification

1. Density-driven transport in the unsaturated zone

- Numerical results published by Mendoza and Frind (1990)
- Density-driven gas flow and contaminant transport

2. Biotransformation of contaminant in the saturated zone

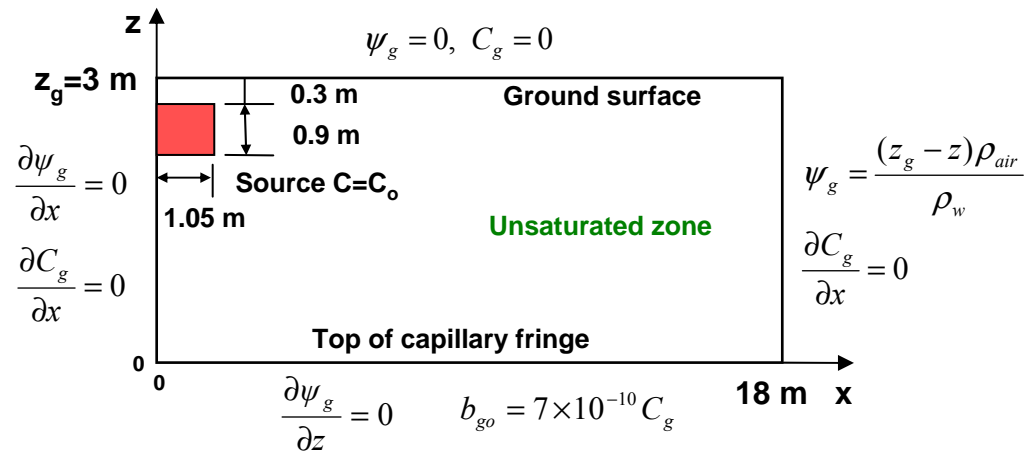
- Analytical solutions in a three-dimensional domain
- Transport of three contaminants

Model verification: 1. Density-driven transport

Gas flow and transport in the unsaturated zone (2D)

Mendoza and Frind (1990)

- Simulation : density-driven transport of generic volatile organic compound (VOC).
- No advective flow of gas phase in the unsaturated zone at $t=0$ (no pressure gradient at gas phase)
- No reaction except equilibrium between water and gas phases
- Constant concentration at source zone



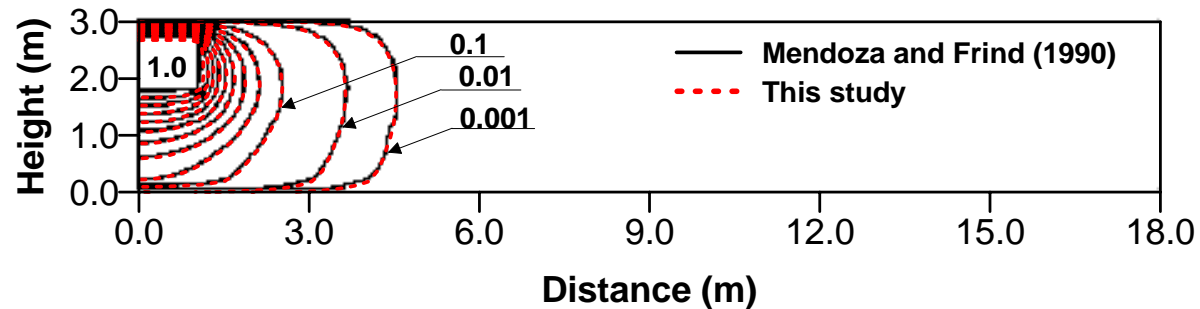
Soil medium	Permeability, k	$1.0 \times 10^{-10} \text{ m}^2$
	Porosity, ϕ	40 %
	Water saturation, s_w	20 %
	Residual water saturation, s_m	20 %
	Pore-size index, λ	1.65 g/cm^3
	Longitudinal dispersivity, α_L	0.15 m
	Transverse dispersivity, α_T	0.0075 m
	Temperature, T	20 °C
Generic VOC	Molecular weight, M_C	100.625 g/mol
	Molecular diffusion coefficient, D^*	$9.0 \times 10^{-6} \text{ m}^2/\text{s}$
	Vapor viscosity, μ_C	$1.0 \times 10^{-5} \text{ Pa s}$
	Henry's constant, H	0.17



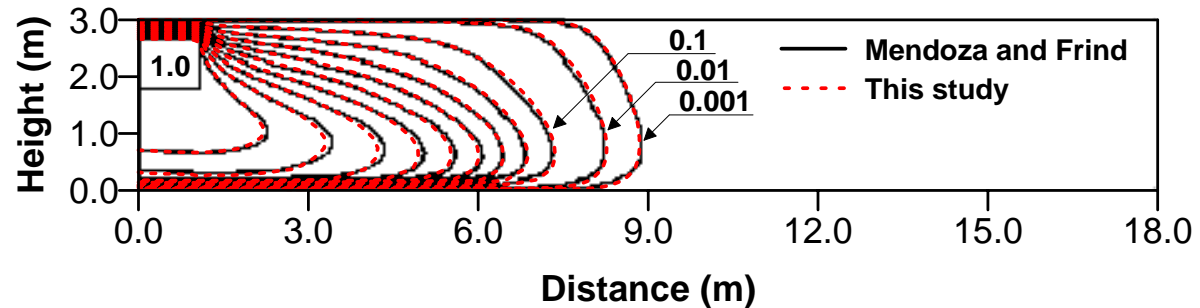
Model verification: 1. Density-driven transport (continued)

- Contaminant transport in gas phase in the unsaturated zone at t=4days

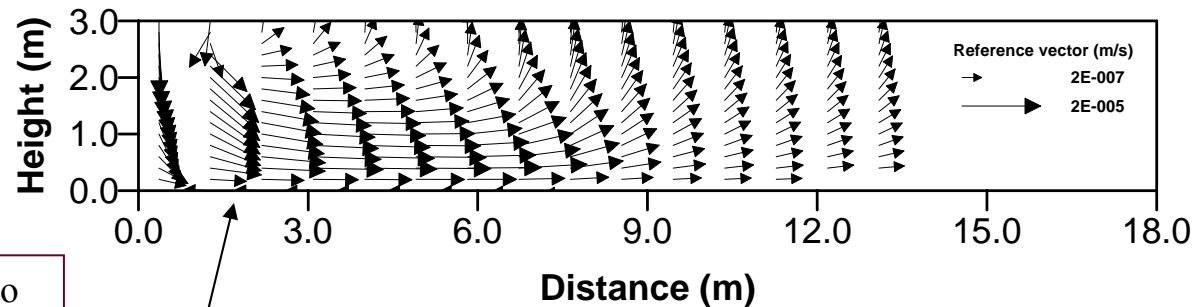
*Molecular diffusion
(No advection)*
Gas density = constant



*Density-driven transport
(Advection + dispersion)*
Gas density = $f(\text{concentration})$



Darcy velocity of gas
Density-driven transport

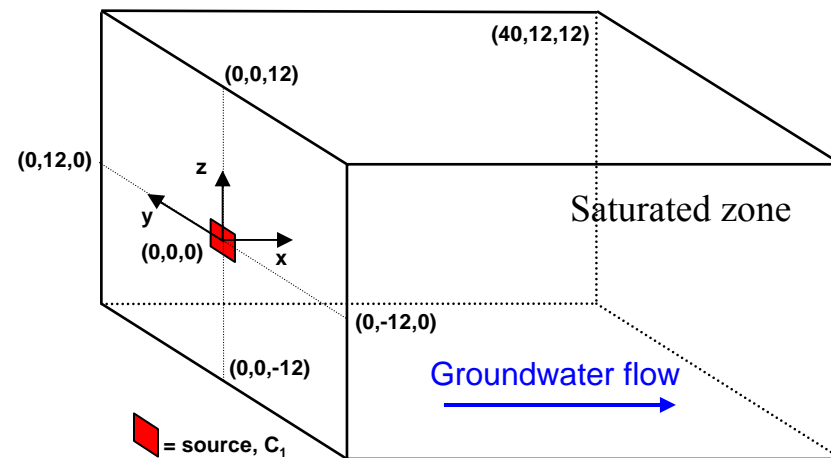
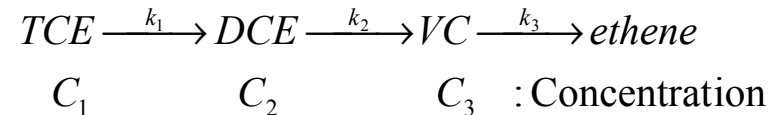


Advection (gas flow) due to density gradient in gas phase

Model verification : 2. Biotransformation

▪ Transport of reactive contaminants in ground water flow (3D)

- Sequential biotransformation (First-order relations)
- Three contaminants
 - For example, TCE, DCE, and VC
- Initial condition
 - At source: $C_1 = 1$, Constant
 - C_2 and $C_3 = 0$. in domain
- Simulation domain
 - Size = 40 m × 24 m × 24 m
 - dx, dy, dz = 0.25 m~1.0 m



TCE: Trichloroethylene
 cDCE: cis-Dichloroethylene
 VC: Vinyl chloride

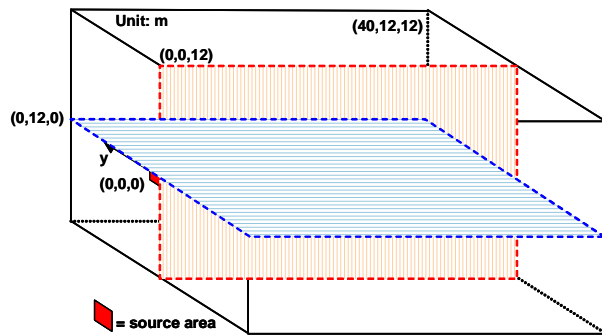
Parameters	Values
q_w , m/d	0.2
k_1, k_2, k_3 , d ⁻¹	0.05, 0.02, 0.01
D_x, D_y, D_z , m ² /d	0.3, 0.3, 0.1

Model verification: 2. Biotransformation (continued)

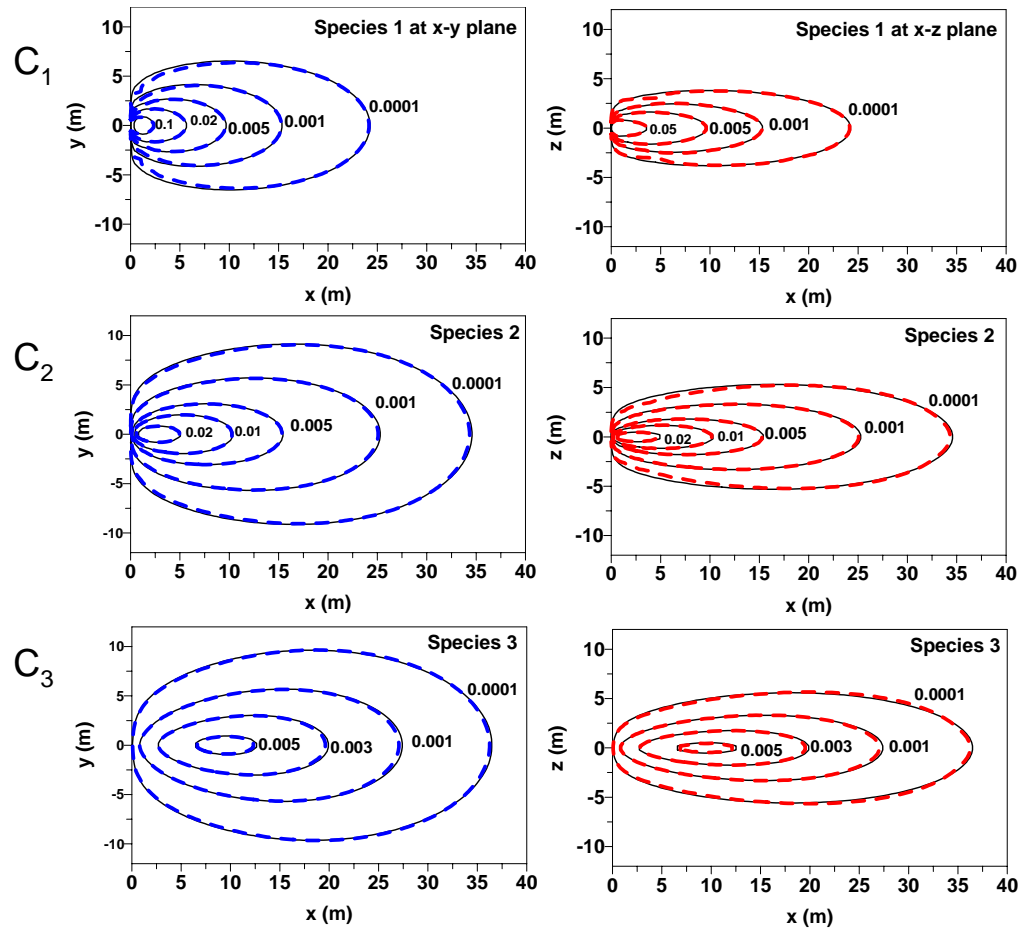
Results : Concentration of reactive contaminants

- Results are compared with analytical solutions

- Analytical solutions (Wexler, 1992)
- - - This study at x - y planes ($z = 0$)
- - - This study at x - z planes ($y = 0$)



t=100 days



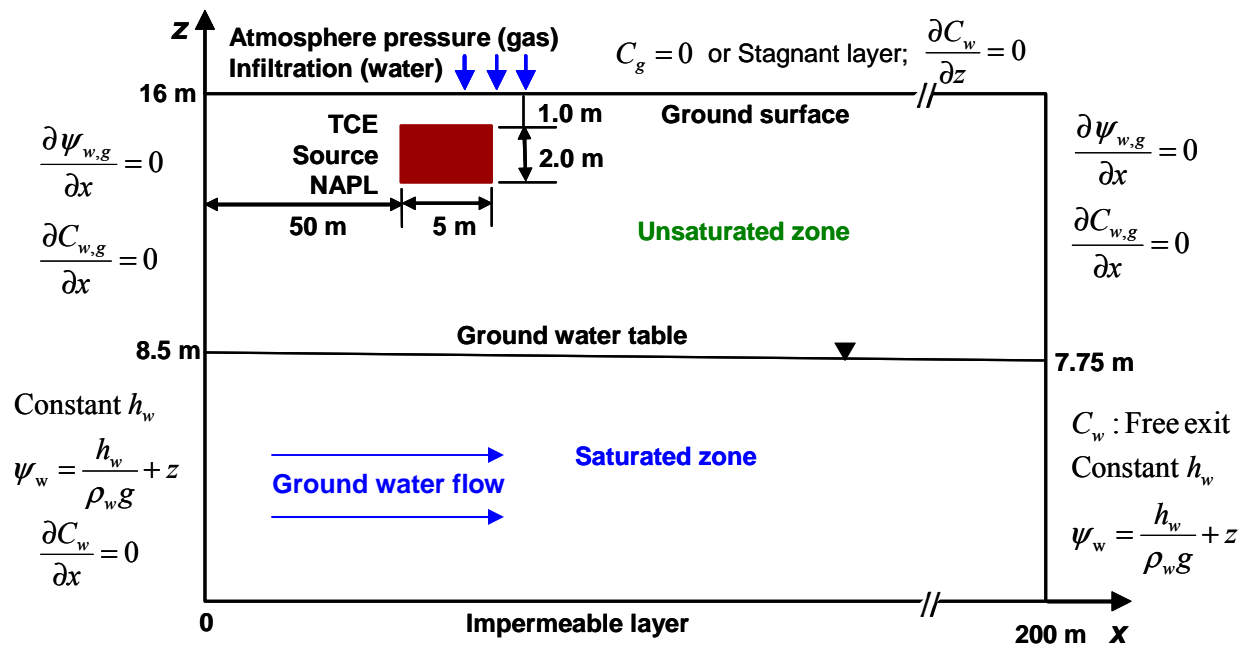


Density-driven transport with Biotransformation

- **The simulation of this study considered:**
 - Both unsaturated and saturated zones in the domain
 - Biotransformation for long-term simulation
 - Quantitative analysis
 - The contributions of important factors to ground water pollution

Modeling domain

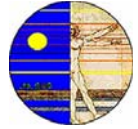
- Simulation of TCE transport in the variably saturated zone
 - Model domain (unsaturated + saturated zone)



TCE source

Initial NAPL saturation = 5%

Initial TCE mass as NAPL = 64.9 kg



Parameters and properties

Parameters	TCE (C ₂ HCl ₃)	cDCE (C ₂ H ₂ Cl ₂)	VC (C ₂ H ₃ Cl)
Molecular weight	131.39	96.94	62.50
Vapor density, kg/m³	5.56	4.10	2.64
Vapor dynamic viscosity, Pa s × 10⁶	9.38	9.29	9.27
Henry constant, dimensionless	0.227	0.097	0.756
Molecular diffusion in air, m²/s × 10⁶	7.87	8.84	10.42
Molecular diffusion in water, m²/s × 10¹⁰	8.206	8.711	10.65
Sorption coefficient, K_{oc}, L/g	0.1	0.049	0.003
Vapor pressure, mmHg	41.27	129.2	2136.30
max. C_g, kg/m³	0.302	0.697	7.434
max. C_w, kg/m³	1.33	7.19	9.83

TCE: Trichloroethylene
 cDCE: cis-Dichloroethylene
 VC: Vinyl chloride

Porous medium

Permeability, k	$1.0 \times 10^{-10} \text{ m}^2$
Porosity, ϕ	0.35
Residual water saturation, s_m	0.
Bulk density, ρ_b	1.6 g/cm ³
Temperature, T	15 °C
Longitudinal dispersivity, α_L	1.0 m
Transverse dispersivity, α_T	0.01 m
Soil organic content, f_{oc}	0.0025
<i>Parameters for the unsaturated zone</i>	
n	2.0
α_{gw}	5.0 m ⁻¹



Simulation scenarios

- **Scenario 1. Diffusion vs. Density-driven transport**

- **Scenario 2. Sequential biological transformations**
: TCE \rightarrow cDCE \rightarrow VC



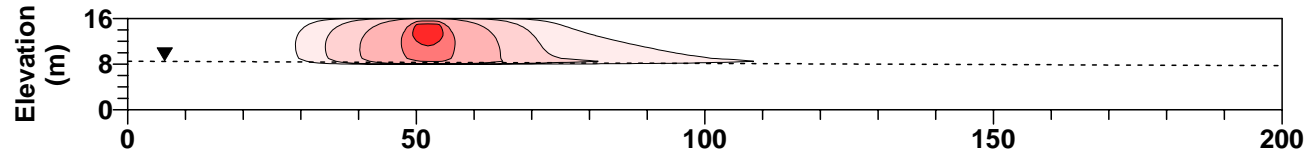
Scenario 1. Diffusion vs. Density-driven transport

Red = Contaminant in gas phase

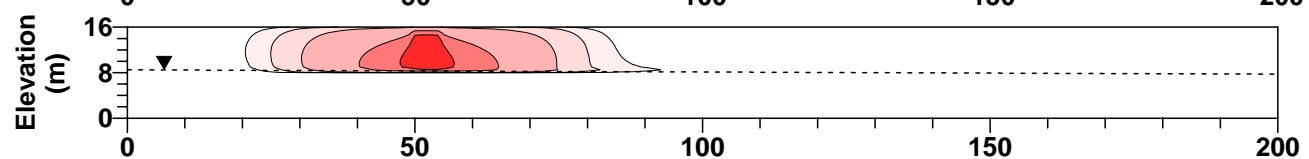
Concentration of TCE in gas phase

- t=100 days

*Diffusion
(no advection)*

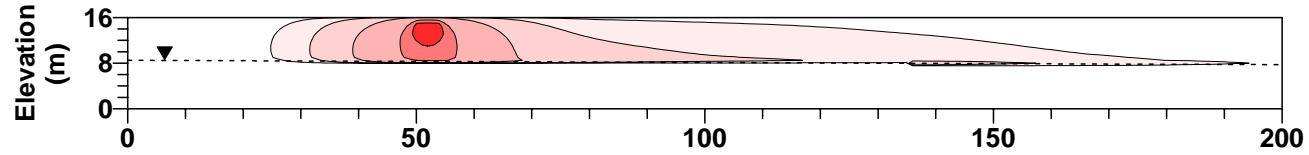


*Density-driven
Transport
(Adv.+Disp.)*

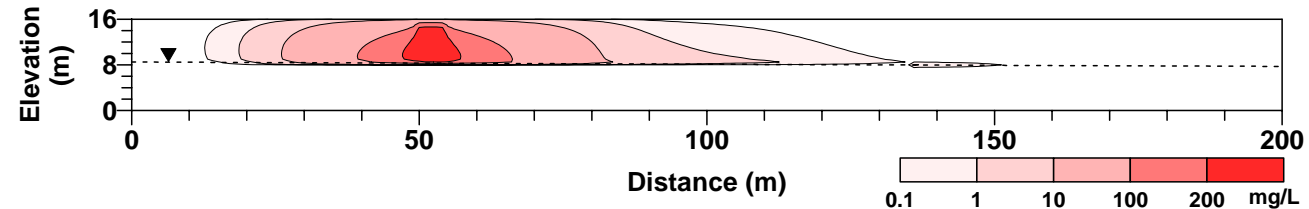


- t=200 days

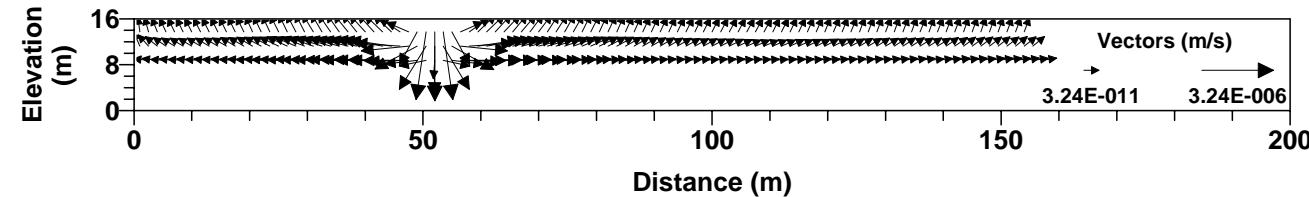
Diffusion



*Density-driven
transport*



Gas phase
Density-driven flow

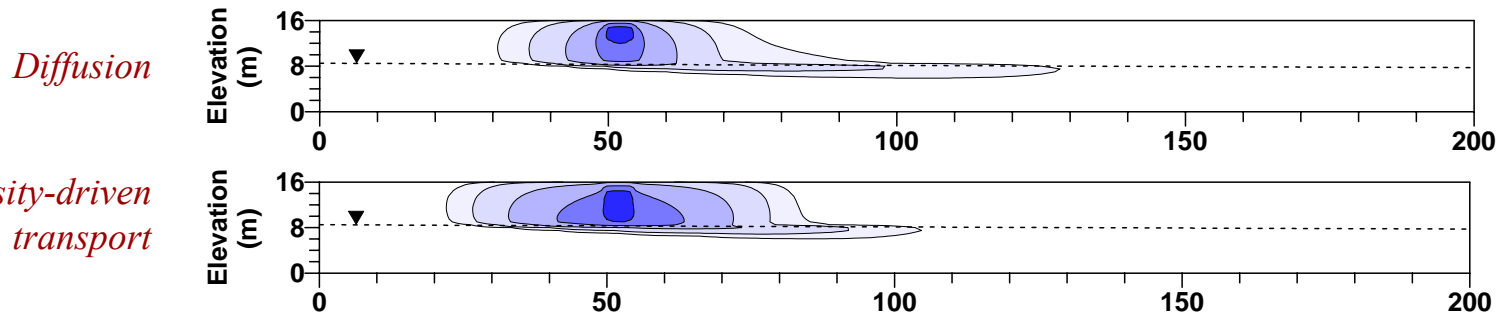


Scenario 1. Diffusion vs. Density-driven transport (continued)

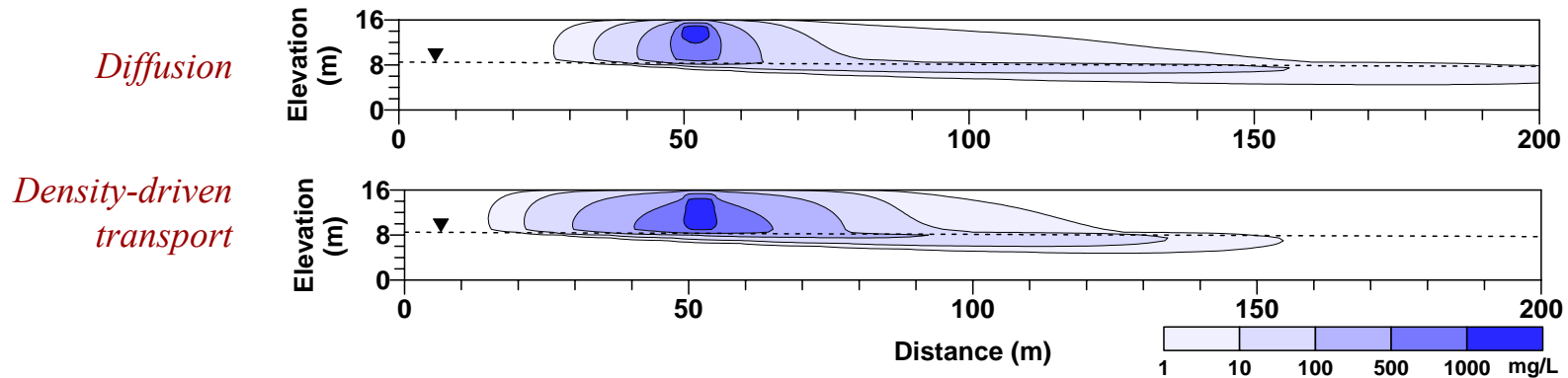
Blue = Contaminant in water phase

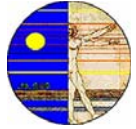
- Concentration of TCE in water phase

- t=100 days



- t=200 days

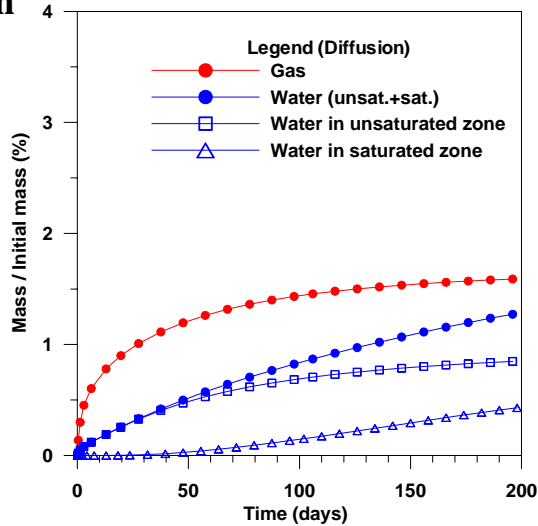




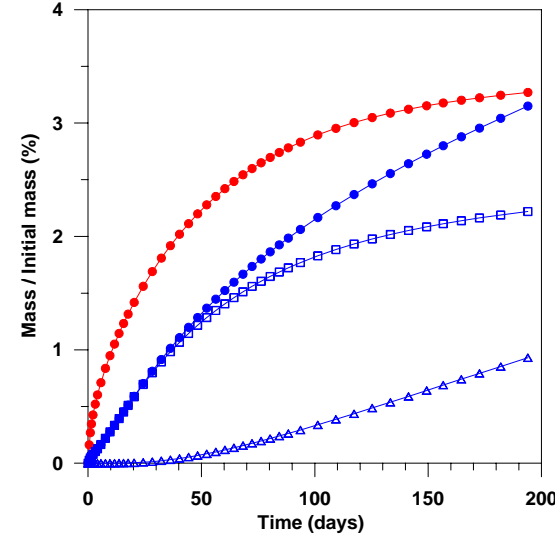
Scenario 1. Diffusion vs. Density-driven transport (continued)

TCE distribution

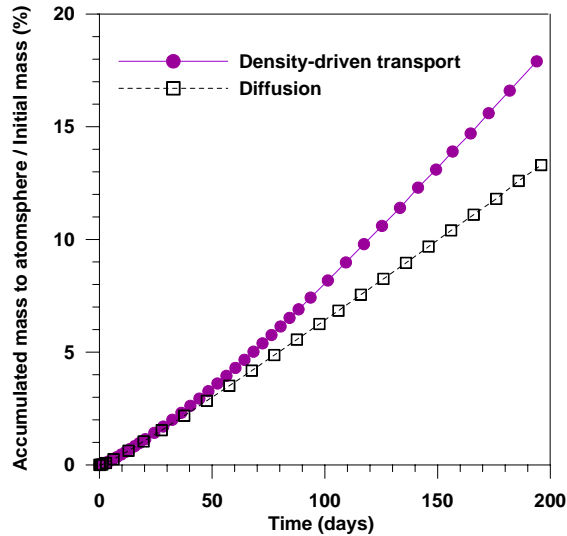
Diffusion



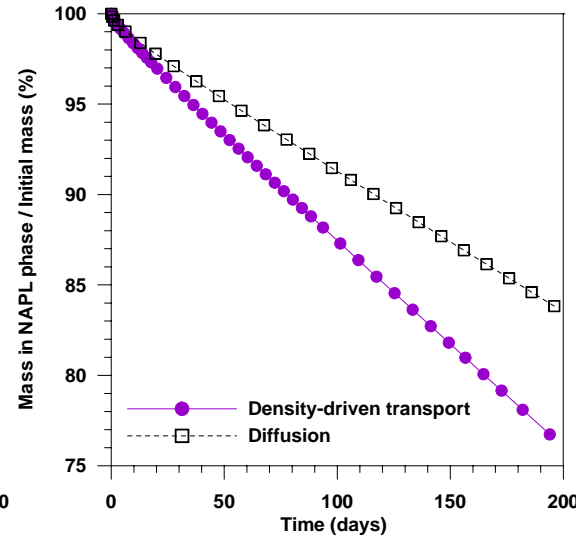
Density-driven transport



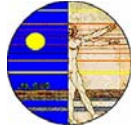
Release of TCE to the atmosphere



Mass reduction in NAPL source



$$\text{Mass balance: } \sum_{Source} TCE_{NAPL}^{Initial\ mass(t=0)} - \sum_{t=0}^t TCE_{Atmosphere} - \sum_{Element} TCE_{Water}^{(t)} - \sum_{Element} TCE_{Gas}^{(t)} - \sum_{Element} TCE_{Soil}^{(t)} - \sum_{Source} TCE_{residual\ NAPL}^{(t)} \approx 0$$



Scenario 2. Biotransformation

- Assumed TCE is dehalogenated via sequential reactions

- It occurred under the anaerobic condition
- First-order reactions

- Effect of biological transformation of TCE

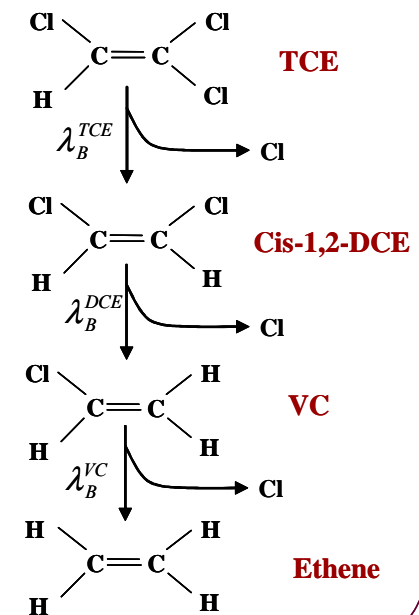
- To investigate byproducts
 - DCE and VC
- First-order bioreaction coefficients (Three cases)

Rate(day ⁻¹)	TCE	DCE	VC
Case I*	3.0×10^{-3}	2.5×10^{-3}	3.8×10^{-3}
Case II	1.5×10^{-3}	1.3×10^{-3}	1.9×10^{-3}
Case III**	1.1×10^{-4}	1.6×10^{-3}	1.0×10^{-3}

*Suna et al. [2001]

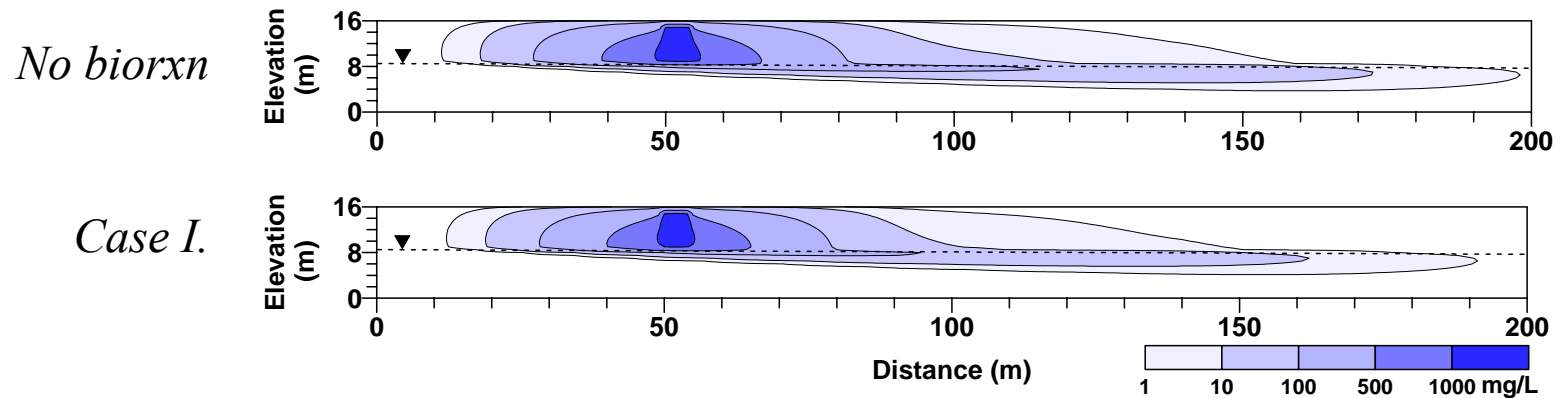
**Clement et al. [2000]

Sequential bioreaction of TCE

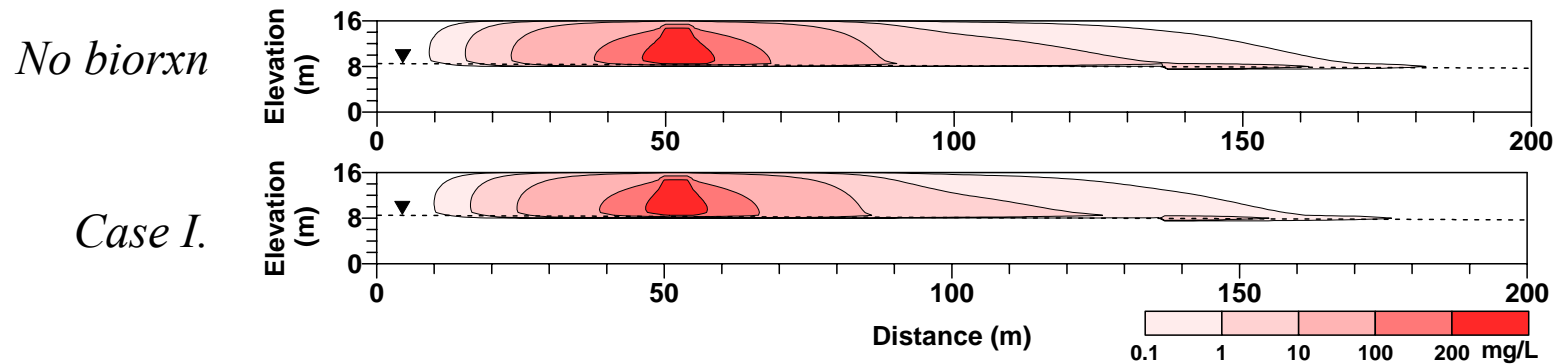


Scenario 2. Biotransformation (continued)

- TCE concentration in water phase at 280 days

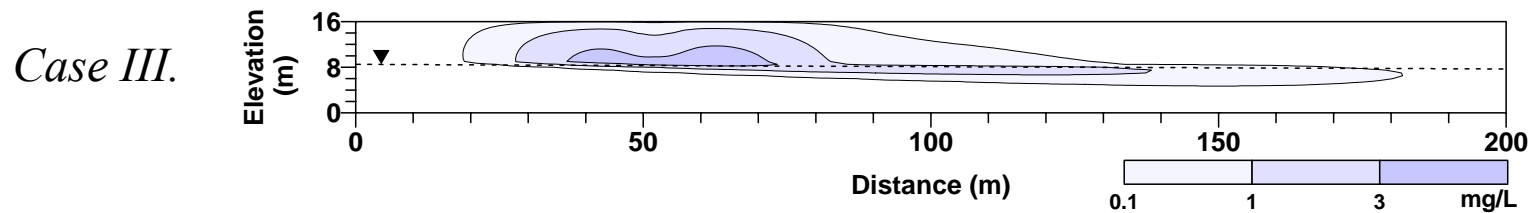
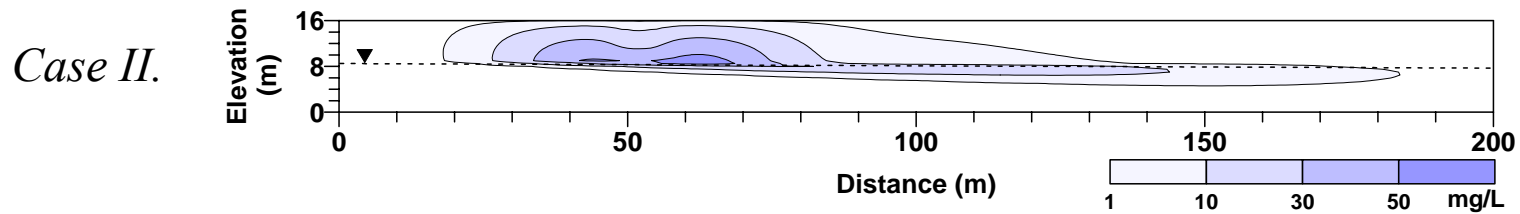
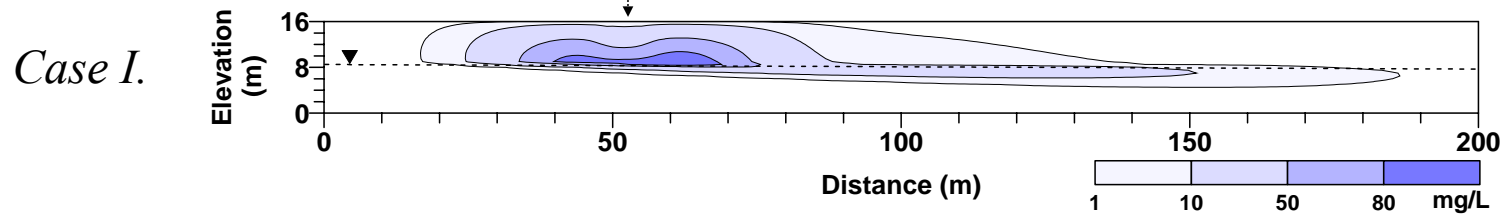
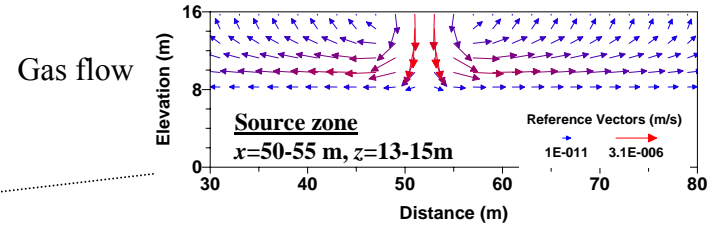


- TCE in gas phase at 280 days



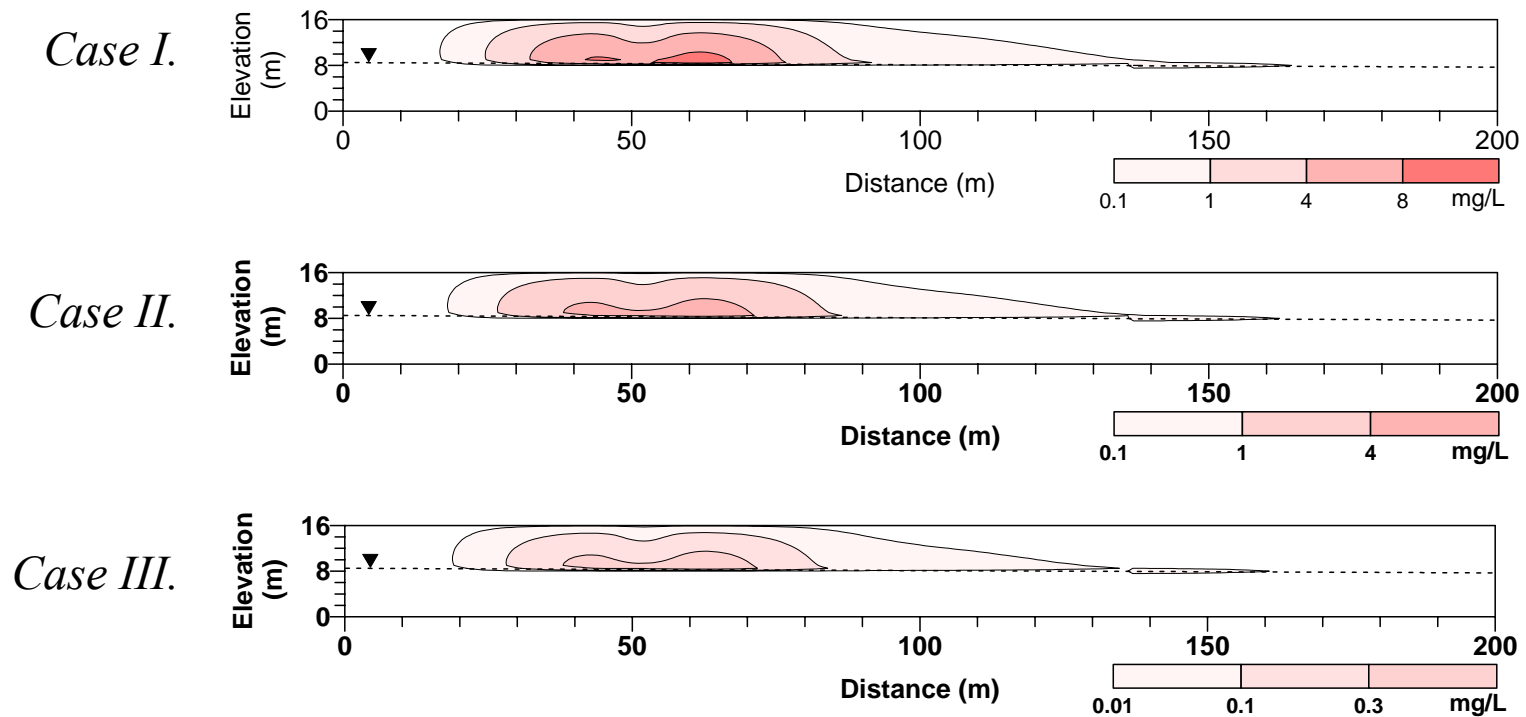
Scenario 2. Biotransformation (continued)

- DCE concentration in water**



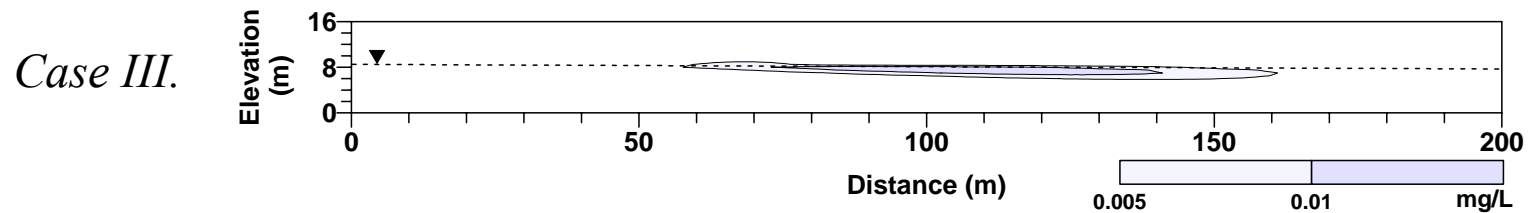
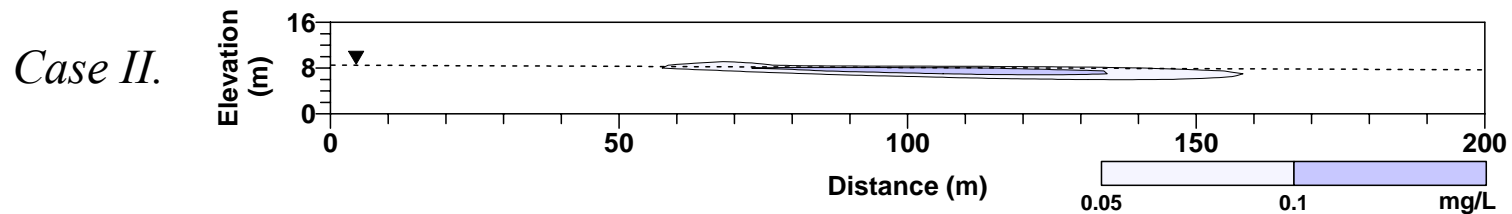
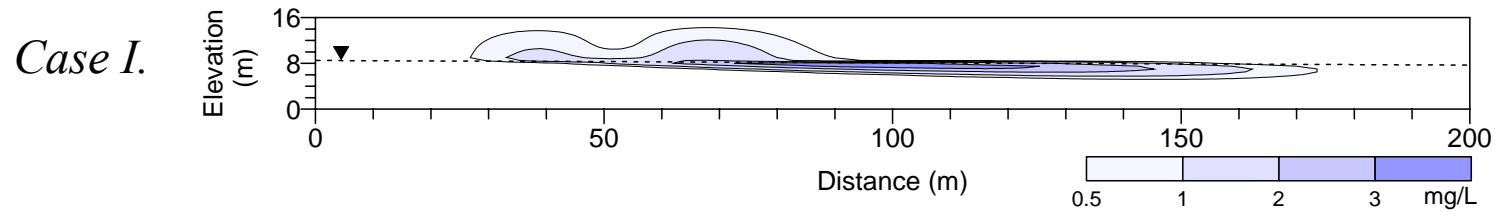
Scenario 2. Biotransformation (continued)

- **DCE concentration in gas phase**

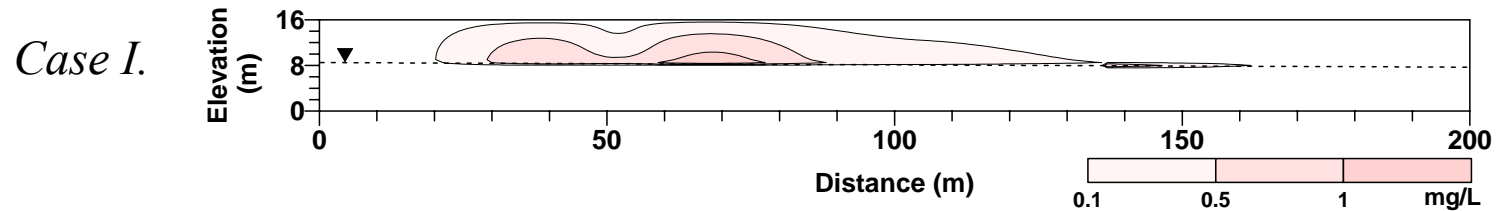


Scenario 2. Biotransformation (continued)

- VC concentration in water phase at t=280 days

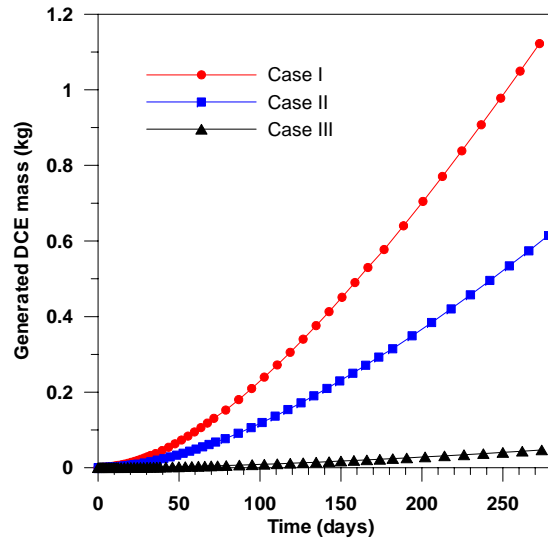


- VC in gas phase

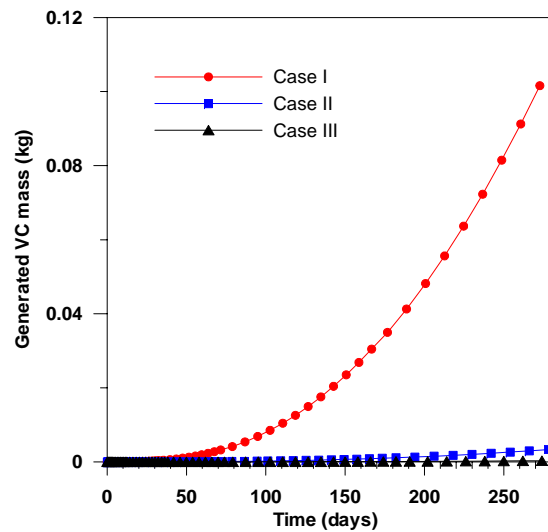


Scenario 2. Biotransformation (continued)

DCE production



VC production



Mobilized TCE at 300 days

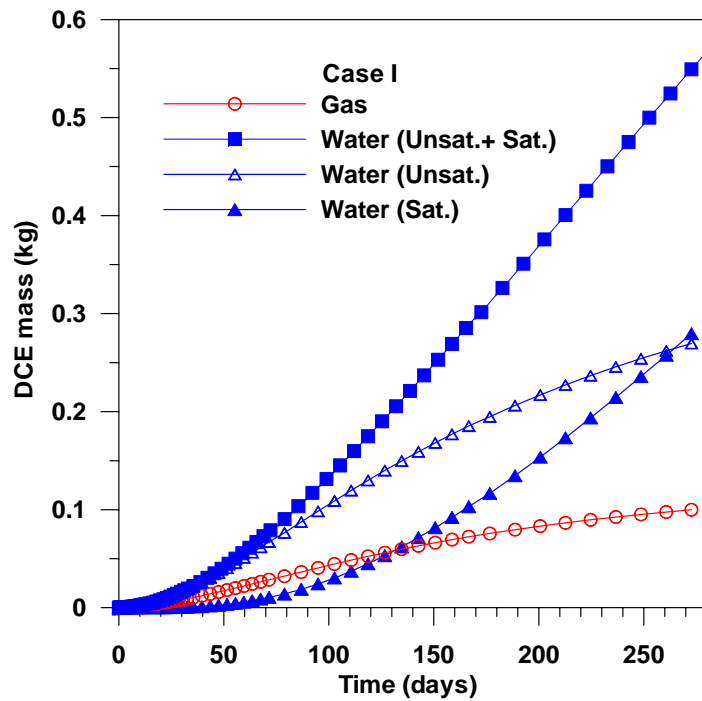
Bioreaction	Mobilized TCE (Dissolved + Vaporized TCE)	
	Mass ¹⁾ , Kg	Ratio, % ²⁾
Case I	21.43	33.55
Case II	21.30	33.35
Case III	21.18	33.16

1) Mobilized TCE (kg) = Dissolved + Vaporized TCE from NAPL source

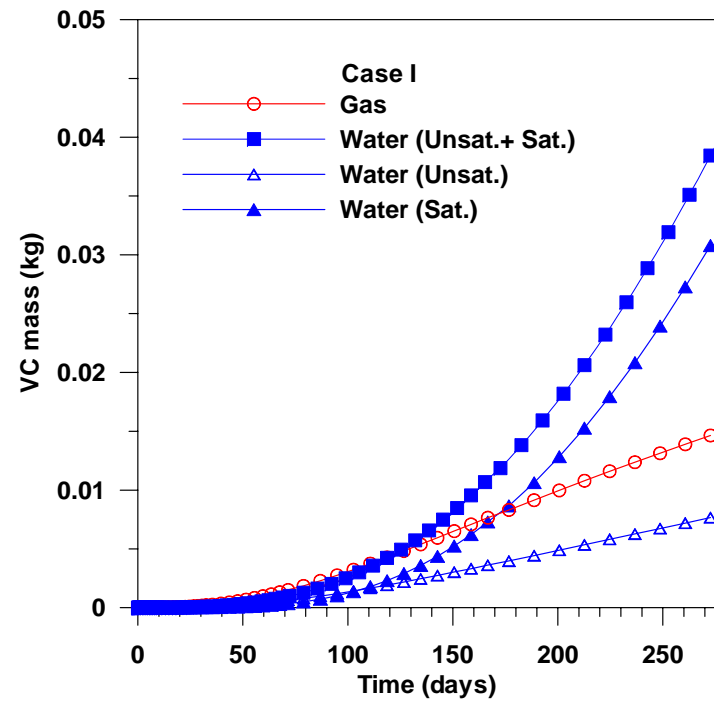
2) % = Mass of mobilized TCE (kg) / Initial TCE mass (kg)

Scenario 4. Biotransformation (continued)

▪ Distribution of DCE and VC



Dichloroethylene



Vinyl chloride



Summary

- **Density-driven transport in the unsaturated zone increases**
 - The spreading of contaminants
 - Contaminant transport to groundwater
 - Contaminant removal to atmosphere

- **Biotransformation**
 - Can generate new toxic contaminants
 - Is important near/in the saturated zone
 - Have an important influence on the distribution of new contaminants
 - Should be considered for long-period simulations.



References

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